Research Article

Magnetic Properties and Microstructure of FeOₓ/Fe/FePt and FeOₓ/FePt Films

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The Fe(6 nm)/FePt film with perpendicular magnetization was deposited on the glass substrate. To study the oxygen diffusion effect on the coupling of Fe/FePt bilayer, the plasma oxidation with 0.5–7% oxygen flow ratio was performed during sputtered part of Fe layer and formed the FeOₓ(3 nm)/Fe(3 nm)/FePt trilayer. Two-step magnetic hysteresis loops were found in trilayer with oxygen flow ratio above 1%. The magnetization in FeOₓ and Fe/FePt layers was decoupled. The moments in FeOₓ layer were first reversed and followed by coupled Fe Pt bilayer. The trilayer was annealed again at 500°C and 800°C for 3 minutes. When the FeOₓ(3 nm)/Fe(3 nm)/FePt trilayer was annealed at 500°C, the layers structure was changed to FeOₓ(6 nm)/FePt bilayer due to oxygen diffusion. The hard-magnetic FeOₓ(6 nm)/FePt film was coupled with single switching field. The FeOₓ/(disordered FePt) layer structure was observed with further annealing at 800°C and presented soft-magnetic loop. In summary, the coupling between soft-magnetic Fe, FeOₓ layer, and hard-magnetic L1₀ FePt layer can be controlled by the oxygen diffusion behavior, and the oxidation of Fe layer was tuned by the annealing temperature. The ordered L1₀ FePt layer was deteriorated by oxygen and became disordered FePt when the annealed temperature was up to 800°C.

1. Introduction

Equiatomic FePt film with anisotropic face-centered tetragonal (fct) L1₀ ordered structure has high magnetocrystalline anisotropy which is the promised material in energy assisted perpendicular magnetic recording. The L1₀ FePt phase was ordered from disordered FePt phase with face-centered cubic (fcc) structure after high temperature annealing [1–6]. The disordered and L1₀ FePt phases show soft- and hard-magnetic properties, respectively. The ordering degree can be changed from 0 (disordered) to 1 (ordered) that depends on the process temperature or condition. The granular structure with columnar grains [7, 8] and well c-axis alignment normal to the film surface with low switching field distribution are required for FePt perpendicular recording media. For c-axis alignment, [001] textured FePt films have been prepared on amorphous glass substrate or MgO underlayer with lower ordering temperatures [9–11]. To write in perpendicular recording media, measure can be taken from energy assisted writing process called heat assisted magnetic recording (HAMR) [12]. To down the writing temperature under fixed writing field, minor adjusting of the intrinsic magnetic anisotropy of FePt film was required. The magnetic anisotropy and the coercivity of FePt film can be tuned by composition and the third element addition [13]. The [001] textured FePt film was not easily formed when target composition far deviated from equal atomic ration. In addition, the magnetic anisotropy was usually diluted by nonmagnetic doping. As a result, it is necessary to have high [001] textured FePt granular film with lower coercivity that was accepted by writing temperature and field. Traditionally, exchange spring media and exchange coupled composite (ECC) media with higher and lower magnetic anisotropic layer were introduced to reduce the writing field requirement and maintain the same thermal stability and grater insensitivity to easy axis [14–20]. The motivations of this work are try (1) to know the oxygen diffusion effects on magnetic properties of FePt film and prove the change of magnetic properties by the second annealing process, (2) to modify the morphology of dewetted FePt film by FeOₓ, and (3) to study the coupling effect via FeOₓ/FePt trilayer structure. The structure of FeOₓ may be cubic FeO, tetragonal γ-Fe₂O₃ (maghemite)
or spinel Fe$_3$O$_4$ (magnetite), with ferrimagnetism. Plasma oxidation and further annealing were proposed and proven to change the magnetic properties and microstructure of FeO$_x$/Fe/FePt films. The [001] textured FeO$_x$/FePt dewetted film with perpendicular magnetization was also observed.

2. Experimental

The FePt film was deposited on a glass substrate at room temperature (RT) by magnetron sputtering. The sputtering system was designed for ultrahigh vacuum with base pressure of $5 \times 10^{-8}$ Torr, and the load-lock system was used to transfer the substrate via prechamber with base pressure of $5 \times 10^{-7}$ Torr. The FePt and Fe targets were used, and a working pressure was fixed at $1.5 \times 10^{-3}$ Torr under high purity argon gas. The total thickness of FePt film is 10 nm, and the chemical composition of the FePt layer was Fe$_{48}$Pt$_{52}$ measured by an energy dispersive spectrometer (EDS) on a single thicker FePt layer. After deposition, the films were annealed under argon atmosphere by using a rapid thermal annealing process (RTP) at 800°C for 3 minutes with heating rate of 10°C/s and formed the L1$_0$ FePt thin film. The Fe layer with thickness of 6 nm was deposited at RT on FePt film and formed the Fe/FePt bilayer. To study the oxygen diffusion effect on the coupling of Fe(6nm)/FePt bilayer, the plasma oxidation with varied oxygen flow ration [$P = P_{O_2}/(P_{O_2} + P_{Ar}) = 0.3, 0.5, 1, 3, 7\%$] was performed during sputtered part of Fe layer and formed the FeO$_x$ (3 nm)/Fe(3 nm)/FePt trilayer. More precisely, the oxygen was introduced during sputtering in half of the Fe layer thickness which is 3 nm. The trilayer was further annealed at 500°C, 800°C and finally formed the FeO$_x$/FePt, FeO$_x$/FePt, FeO$_x$/Fe(3 nm)/FePt (disordered FePt), respectively. The thickness of each layer or sputtering rate of each material was measured by atomic force microscopy. The crystal structure of the samples was identified using a standard X-ray diffraction (XRD) technique (BRUKER, D8 Discover). The film microstructure was observed by scanning electron microscopy (JEOL JSM-6700F) and atomic force microscopy (DI 3100). The surface chemical property was measured by X-ray photoelectron spectroscopy (XPS) PHI5000Versa-Probe. Magnetic hysteresis loops were measured at room temperature using a vibration sample magnetometer (VSM Lakeshore 7400) with a maximum magnetic field of 2T. The magnetic field was applied to be parallel and perpendicular to film surface to obtain in-plane and out-of-plane hysteresis loops, respectively.

3. Results and Discussion

Figures 1(a)–1(f) show standard XRD patterns of FePt single layer, Fe(6 nm)/FePt bilayer, and FeO$_x$/Fe/FePt films. In Figure 1(a), the FePt single layer was ordered in L1$_0$ phase, and the (001) superlattice diffraction peak and the (002) fundamental reflection are clearly observed. The XRD profiles suggest that the L1$_0$ FePt crystal has a [001] texture. Figures 1(b)–1(f) show the XRD patterns of Fe(6 nm)/FePt bilayer and FeO$_x$(P)/Fe/FePt trilayer ($P = 0.5, 1, 3, \text{ and } 7\%$). In Figures 1(b)–1(f), the L1$_0$ FePt films also show strong (001) and (002) peaks that were not smeared out by Fe and FeO$_x$ deposited at RT. The peaks of Fe and FeO$_x$ were not found in standard X-ray diffraction (XRD) due to strong FePt (001) texturing. The standard X-ray peaks of Fe (JCPDS 89-7194), FeO (JCPDS 89-7100), Fe$_3$O$_4$ (JCPDS 89-8104), and FeO$_x$ (JCPDS 89-6466) were marked in Figures 1 and 2. The ordering parameter, $S$, was calculated from $I_{(002)}^*/I_{(001)}^*$ ratio [21]. The $I_{(002)}^*/I_{(001)}^*$ value, for example, 0.4915 in L1$_0$ FePt film, was obtained after considering all the corrected factors in the XRD data. Fully-ordered L1$_0$ FePt X-ray diffraction peak intensity is given by $I_{(002)}^*/I_{(001)}^* = ([F]^2 \times \text{LPDA})_{002}/([F]^2 \times \text{LPDA})_{001}$ [21]. For L1$_0$ FePt, the structure factor $F$ is $F_{hid} = f_{Fe}(1 + e^{i(h+k)}) + f_{Pt}(e^{i(h+k+l)} + e^{-i(h+k+l)})$, and $f$ is atomic scattering factor [21]. The...
Figure 3: In-plane and out-of-plane magnetic hysteresis loops of (a) FePt single layer, (b) Fe(6 nm)/FePt bilayer, and FeOx(P)/Fe/FePt films with different oxygen flow rates; P = (c) 0.5%, (d) 1%, (e) 3%, and (f) 7%.
Figure 4: In-plane and out-of-plane magnetic hysteresis loops of FeO$_x$(P)/Fe/FePt trilayer further annealed at 500°C with $P = (a)$ 1%, (b) 3%, and (c) 7% and 800°C with $P = (d)$ 1%, (e) 3%, and (f) 7%. 
Lorentz factor \( (L) \) is \( [1/\sin^2(\theta)\cos(\theta)] \), and the polarization factor \( (P) \) is \( (1 + \cos^2(2\theta)) \). The temperature factor \( (D) \) is \( e^{-\frac{2M}{\sin(\theta)}} \), and absorption factor \( (A) \) is \( 1 - \exp(-2\mu t/\sin(\theta)) \). The average mass absorption coefficient was estimated as \( \mu = [\mu_{Fe} \times wt\%Fe + \mu_{Pt} \times wt\%Pt] \times [X_{Fe}\rho_{Fe} + X_{Pt}\rho_{Pt}] \) [21], and the value is 3323.8 in this study. Here, \( X_{Fe} \), \( X_{Pt} \), \( \rho_{Fe} \), and \( \rho_{Pt} \) are the atomic fraction and density of Fe and Pt, respectively. The values for \( \mu_{Fe} \) and \( \mu_{Pt} \) are tabulated [21]. \( I_{\theta001}^{(001)} \) and \( I_{\theta002}^{(002)} \) mean the theoretical integrated intensity of X-ray diffraction peaks, and the \( I_{\theta001}^{(001)} \) and \( I_{\theta002}^{(002)} \) are the integrated peak intensity from the experimental results for a partially ordered film. In Figure 1(a), the ordering parameter \( S \) estimated from \( (I_{\theta002}^{(002)}/I_{\theta001}^{(001)})^{1/2} \) is 0.75, and the \( S \) value in Figures 1(c)–1(f) is 0.71–0.72.

Figures 2(a)–2(f) show standard XRD patterns of FeO\(_x\)/Fe/FePt trilayer further annealed at 500° C and 800° C, respectively. In Figures 2(a)–2(c), the FeO\(_x\)/Fe/FePt trilayer was further annealed at 500° C and became FeO\(_x\)/FePt bilayer. In addition to the widthness (full width of half maximum (FWHM)) of (001) and (002) peaks in Figure 2(b), the \( I_{\theta001}^{(001)} \) FePt still shows strong (001) texture. It means that the \( I_{\theta001}^{(001)} \) FePt layer was not influenced almost by further being annealed at 500° C. In Figures 2(d)–2(f), the FeO\(_x\)/Fe/FePt trilayer was further annealed at 800° C, and the (001) texture was almost diminished, and (200) dispersed FePt peak was observed. It is suggested that the oxygen was diffused into the \( I_{\theta001}^{(001)} \) FePt layer and the layer, structure was changed to FeO\(_x\)/(disordered FePt). The FePt (111) peak was not found in standard XRD patterns in Figures 1 and 2 due to strong FePt (001) texture. The ordered or disordered FePt (111) peak was indexed in grazing incident X-ray diffraction (GID). The nonsymmetric GID was used to measure the grains in nonoriented area in textured film.

Figure 3 shows in-plane and out-of-plane magnetic hysteresis loops of FePt single layer, Fe(6 nm)/FePt bilayer, and FeO\(_x\)(3 nm)/Fe(3 nm)/FePt trilayer. In Figure 3(a), the FePt single layer shows high perpendicular coercivity \( (H_c) \) of 15.3 kOe. The component in in-plane magnetization is nearly zero and shows the linear loop. In Figure 3(b), the magnetization was increased, and the \( H_c \) was reduced to 11.8 kOe in soft/hard exchange coupled Fe/FePt bilayer. Figures 3(c)–3(f) show the loops of FeO\(_x\)/Fe/FePt trilayer with different oxygen flow rates \( [P = P_{O_2}/(P_{O_2} + P_{Ar})] = 0.5, 1, 3, 7\% \). In Figure 3(c), there is just a minor shoulder in the loop of FeO\(_x\)/Fe/FePt film with low oxygen flow rates \( (P = 0.5\%) \). When the oxygen flow ration was increased to 1, 3, and 7%, the step or shoulder was found in magnetization curve in Figures 3(d)–3(f). The FeO\(_x\) layer was decoupled to the Fe/FePt bilayer and reversed the magnetization previously. The magnetization in FeO\(_x\) layer was decreased as the negative applied field increased, and, up to the critical value, the magnetization was not changed with increased field. The critical value was defined as the \( H_c \) of FeO\(_x\) layer. In Figures 3(d)–3(f), the \( H_c \) of the FeO\(_x\) layer is 1.37, 1.66, and 1.09 kOe, and the \( H_c \) values of Fe(3 nm)/FePt bilayer are 13.7, 13.7, and 13.4 kOe, respectively. The \( H_c \) value of Fe(3 nm)/FePt is between FePt single layer in Figure 3(a) and Fe(6 nm)/FePt in Figure 3(b).

The samples in Figures 3(d)–3(f) are the same to Figures 4(a)–4(c). The trilayers (samples in Figures 3(d)–3(f)) were further annealed at 500° C and 800° C, respectively. Figure 4 shows in-plane and out-of-plane magnetic hysteresis loops of FeO\(_x\)(3 nm)/Fe(3 nm)/FePt trilayer. Figures 4(a)–4(c) show the loops of FeO\(_x\)(3 nm)/Fe(3 nm)/FePt films annealed at 500° C with oxygen flow ration of 1%, 3%, and 7%, respectively. The shoulders appearing in the loops in Figures 3(d)–3(f) disappeared after further annealing, and the loops present single magnetization reversal process. It suggested that the Fe layer was further oxidative, and the layer structure was changed to exchange coupled FeO\(_x\)/FePt film. In Figures 4(a)–4(c), the out-of-plane \( H_c \) were 14.4, 10.4, and 14.8 kOe which are similar to the measured \( H_c \) in Figures 3(d)–3(f). Figures 4(d)–4(f) show the loops of FeO\(_x\)(3 nm)/Fe(3 nm)/FePt films annealed at 800° C with oxygen flow ration of 1%, 3%, and 7%, respectively, and the soft-magnetic loops were obtained. The oxygen was diffused into FePt layer and disordered the \( I_{\theta001}^{(001)} \) phase, and the layer structure was changed to FeO\(_x\)/(disordered FePt). In summary, the interlayer coupling and magnetic anisotropy were tuned by the kinetic diffusion behavior of oxygen driven by temperature.

To understand the oxidation state of Fe and FeO\(_x\) (oxygen flow ration of 3%) layer in FeO\(_x\)(3 nm)/Fe(3 nm)/FePt film, XPS were performed on samples with different depth profiles. Figures 5(a)–5(c) show the Fe-2p X-ray photoelectron spectra of FeO\(_x\)(3 nm)/Fe(3 nm)/FePt films at the depth position below 3 nm from film surface: (a) RT, (b) 500° C, and (c) 800° C, and at the depth position within 3 nm from film surface: (d) RT, (e) 500° C, and (f) 800° C.

![Figure 5: The Fe-2p X-ray photoelectron spectra of FeO\(_x\)(3 nm)/Fe(3 nm)/FePt films at the depth position below 3 nm from film surface: (a) RT, (b) 500° C, and (c) 800° C, and at the depth position within 3 nm from film surface: (d) RT, (e) 500° C, and (f) 800° C.](image-url)
prove the formation of iron in the Fe$^{2+}$ oxidation state that was FeO, and the binding energy of the Fe-2p$_{3/2}$ core level was shifted around to 709.6 eV [23]. In Figure 5(c), the FeO$_x$(3 nm)/Fe(3 nm)/FePt films were further annealed at 800°C, and the peak was smeared out in the iron-oxide area. Figures 5(d)–5(f) show the Fe-2p X-ray photoelectron spectra of FeO$_x$(3 nm)/Fe(3 nm)/FePt films at the depth within 3 nm from film surface. In Figures 5(d)–5(e), the FeO$_x$ layer before and after further annealing at 500°C was in the Fe$^{2+}$ oxidation state that was FeO, and the binding energy of the Fe-2p$_{3/2}$ core level was shifted around to 709.3 eV. In Figure 5(f), the peak was also smeared out in the iron-oxide area when annealed at 800°C. In summary, first, further annealed at 500°C, the Fe (3 nm) may oxidize or mix with FePt film and form FeO$_x$/FePt layer structure. Second, further annealed at 800°C, the oxygen was diffused into the FePt lattice, deteriorated the ordering degree, and formed the FeO$_x$/FePt film.

Figure 6 shows the scanning electron microscopy (SEM) image of Fe(6 nm)/FePt and FeO$_x$(3 nm)/Fe(3 nm)/FePt films with oxygen flow ration of 7%. In Figure 6(a), the annealed Fe(6 nm)/FePt film was dewetted in the network structure, and the dewetted area was 44%. When the Fe layer was partially plasma oxidized and formed the FeO$_x$(3 nm)/Fe(3 nm)/FePt layer structure, the dewetted area was reduced to 30% as shown in Figure 6(b). The FeO$_x$ has lower surface energy than FePt and was easy to wet on the glass substrate. As a result, the dewetted area was reduced around 14%. In Figures 6(c) and 6(d), the FeO$_x$(3 nm)/Fe(3 nm)/FePt film was furthered annealed at 500°C, 800°C, and the dewetted area was 42%, 55%, respectively. Due to large difference of surface energy between FePt and glass substrate, the dewetted area was increased again with annealing temperature.

Figure 7 shows the surface roughness of Fe/FePt and FeO$_x$/Fe/FePt film with oxygen flow ration of 7% measured by atomic force microscopy (AFM). In Figure 7(a), the average surface roughness of Fe/FePt film was 7.1 nm. In Figures 7(b) and 7(c), when the FeO$_x$ layer was capped on Fe/FePt film at RT without annealing or with further annealing at 500°C, the average surface roughness became 4.9 nm and 3.2 nm, respectively. In Figure 7(d), the surface roughness was up to 13 nm when the annealed temperature was high to 800°C. In summary, the dewetted area in Fe/FePt film was partially covered by FeO$_x$ layer deposited at RT but aggregated into small area again with further annealing at high temperature. The surface roughness shows the same tendency to the dewetted area. The FePt film with thickness of 10 nm was dewetted due to strain release when the ordering (rapid phase transformation) was complete at 800°C. The dewetted area and surface roughness were increased with annealing temperature, and the critical dewetted temperature is 650°C in this study. After annealing, the sample was cooled down naturally in the argon atmosphere in RTP system. Cooling rate was not changed in this experiment and will be check in the future.
4. Conclusions

The plasma oxidation was performed during sputtered part of Fe layer in Fe/FePt film and formed the FeO\textsubscript{x}(3 nm)/Fe(3 nm)/FePt trilayer. The magnetization in FeO\textsubscript{x} and Fe/FePt layers was decoupled and shown the shoulders in hysteresis loops. When the FeO\textsubscript{x}(3 nm)/Fe(3 nm)/FePt trilayer was further annealed at 500°C, the layers structure was changed to FeO\textsubscript{x}(6 nm)/FePt bilayer due to oxygen diffusion. The hard-magnetic FeO\textsubscript{x}(6 nm)/FePt film was coupled with single switching field. The coupling between soft-magnetic Fe, FeO\textsubscript{x} layer, and hard-magnetic L\textsubscript{10} FePt layer was tuned by two-stage annealing process.

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References


